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It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system. Third law of thermodynamics It states that absolute zero cannot be reached.

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16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to cold objects. • Heat is the transfer of thermal energy from one object to another because of a temperature difference. Temperature is related to the average kinetic energy of the particles in an object due to their random motions through space.

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Chapter 16 Thermal Energy and Heat

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Chapter 16 Thermal Energy and Heat
Section 16.1 Thermal Energy and Matter
(pages 474-478) This section defines
heat and describes how work,
temperature, and thermal energy are
related to heat. Thermal expansion and

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contraction of materials is discussed, and uses of a calorimeter are explained.

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