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contents:  
thermodynamics .  
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properties and state of pure substances.

chapter 02: work and heat. chapter 03:

energy and the first law of

thermodynamics.

chapter 04: entropy and the second law of thermodynamics.

chapter 05:

irreversibility and availability

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Solutions Homework 2:  
Textbook problems 2.1,  
2.3, 2.4, 2.5 Homework  
2 Solutions Homework  
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Problem Set J Problem

Solution Problem Set K

Problem Solution ...

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## Engineering **10.213-Problem Sets**

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ermodynamics:-Proble

m 1:-A container holds  
a mixture of three

nonreacting gases:  $n_1$

moles of the first gas

with molar specific

heat at constant

volume  $C_{v1}$ , and so on.

Find the molar specific

heat at constant

volume of the mixture,

in terms of the molar

specific heats and

quantities of the three

separate gases.



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CONCEPTS | 5 1.

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PRELIMINARY

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**CONCEPTS - THE LANGUAGE OF THERMODYNAMICS** In order to accurately and precisely discuss various aspects of thermodynamics, it is essential to have a well-defined vernacular. As such, a list of some foundational concepts and their definitions are shown

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Problem Set 1 : 7: 1st

and 2nd Laws Open

and Closed Systems:

Problem Set 2 : 11:

Availability and

Calculus of

Thermodynamics:

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Problem Set 3 : 17:  
Legendre Transforms,  
Phase Rule, Equilibrium  
and Stability Criteria:  
Problem Set 4 : 19:  
Pure Fluid Properties  
Coupled to 1st and 2nd  
Laws, Fugacity,  
Activity, Gibbs-Duhem:  
Problem Set 5

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thermodynamics, we  
can predict the amount

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of energy needed to change a system from an equilibrium state to another. For example it will take about 75 kJ to change 1 kg of air at

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1-7

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Since different units are used to measure energy in its different forms, solving

thermodynamics

problems requires a knowledge of the

conversion ratios (for

example, the ratio for

conversion of foot

pounds, which are

units of work energy,

into BTUs, which are

units of heat energy).

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**Problem Solving in**

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Wednesdays 3-5 ERC  
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Challa Help Session  
Wednesdays 6-9pm  
ERC 405) Professor  
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## THERMODYNAMICS PRACTICE PROBLEMS FOR NON-TECHNICAL MAJORS

Thermodynamic  
Properties 1. If an  
object has a weight of  
10 lbf on the moon,  
what would the same  
object weigh on  
Jupiter? Jupiter 22Moon  
c ft ft lbf-ft g = 75 g  
= 5.4 g = 32 sec sec lbf-  
sec<sup>2</sup> c moon cmoon  
Jupiter Jupiter c mg  
Wg10×32 W = m = =

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59.26 lb gg5.4 mg

59.26×75 W = 139 lbf

g32 ...

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## **Thermodynamic Properties**

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including chemical process design, materials processing, and cellular processes.

Course Objectives: In

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this course, students learn how to apply knowledge of the laws of thermodynamics, chemistry, physics, and engineering to analyze and solve physical and chemical problems encountered in chemical and biochemical engineering.

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