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percent solution by
volume or by mass. A
10% of NaCl solution
by mass has ten grams
of sodium chloride
dissolved in 100 ml of
solution. Weigh 10g of
sodium chloride. Pour it

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into a graduated
cylinder or volumetric
flask containing about
80ml of water. Once
the sodium chloride
has dissolved
completely (swirl the
flask gently if
necessary), add water
to bring the volume up
to the ...

**How do you make a
10 percent solution?**

| Socratic

You have 15mL of a 2%
w/v solution. In order

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to reduce its strength
to 0.4% w/v, how much
solvent should you
add? Q5 What weight
of base should be
added to 12g of a 20%
w/w concentration in
order to produce a 12%
w/w concentration?

Answers Q1 = 950mL.
Q2 = 1800mL. Q3 =
40g. Q4 = 60mL. Q5 =
8g Step 1: Use $c_1 \times v_1$
 $= c_2 \times v_2$ Step 2:
Transpose ...

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